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कुल पृष्ठ संख्या-24 (कवर पेज सहित)

क्रम संख्या



माध्यमिक शिक्षा बोर्ड, राजस्थान, अजमेर

माध्यमिक परीक्षा

(परीक्षार्थी द्वारा स्वयं भरा जाना चाहिये)

Blank space for student details.

प्रश्नवार प्राप्तांकों की सारणी (परीक्षक के उपयोग हेतु)

प्रश्नों की क्रम संख्या	प्राप्तांक	प्रश्नों की क्रम संख्या	प्राप्तांक
1	11	19	3
2	6	20	3
3	10½	21	4
4	2	22	4
5	2	23	4
6	2	24	
7	2	25	
8	2	26	
9	2	27	
10	2	28	
11	2	29	
12	2	30	
13	2	31	
14	2	योग	77½
15	2	प्राप्त अंकों का कुल योग (Round off)	
16	2	अंकों में	शब्दों में
17	3	78	Seventy Eight
18	3		

नोट :- परीक्षार्थी उपरोक्त के अतिरिक्त उत्तर पुस्तिका के अन्य किसी भी भाग में अपना नामांक नहीं लिखें।

माध्यम - हिन्दी अंग्रेजी

विषय Science

परीक्षा का दिन Wednesday

दिनांक 29-03-2023

नोट :- परीक्षार्थी के लिए आवश्यक निर्देश इस पृष्ठ के पिछले भाग पर उल्लेखित हैं। जिन्हें सावधानी पूर्वक पढ़ें लें व पालना अवश्य करें।

परीक्षक हेतु निर्देश :- (1) परीक्षक को उपरोक्त सारणी अनुसार प्राप्तांक भरना अनिवार्य है, अन्यथा नियमानुसार दंडित किया जायेगा।

(2) परीक्षक उत्तर पुस्तिका के अन्दर के पृष्ठों के बायीं ओर निर्धारित कॉलम में लाल इंक से अंक प्रदत्त करें।

(3) कुल योग भिन्न में प्राप्त होने पर उसे पूर्णांक में ही परिवर्तित कर अंकित करें (उदाहरणार्थ : 15 ¼ को 16, 17 ½ को 18, 19 ¾ को 20)

परीक्षक के हस्ताक्षर

G.M. संकेतांक 52482

प्रमाणित किया जाता है कि इस उत्तर पुस्तिका के निर्माण में 58 जी.एस.एम. ईको मैपलियो कागज ही उपयोग में लिया गया है। 175/2023



Section - A

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i	ii	iii	iv	v	vi	vii	viii	ix	x	xi	xii
Ans. C	C	C	A	A	D	C	B	D	B	A	B

2. Fill in the blanks -

(6) (i) The saliva contains an enzyme called salivary amylase.

(ii) Advantage of vegetative propagation is that all plants produced by this are genetically similar to the parent plant.

(iii) Substances that are decomposed by biological processes are called biodegradable.

(iv) Like poles of magnet repel with each other.

(v) Chemical name of acid present in vinegar is ethanoic acid.

(vi) Number of elements present in third period of modern periodic table are eight (8).



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103 3. very short answer type question -

A(i) The mechanism which regulates the action of a hormone is gland. (11)

A(ii) An organism that exhibits multiple fission is Plasmodium. (11)

A(iii) Disadvantage of using fossil fuels is that on burning fossil fuels they release oxides of carbon, nitrogen and sulphur which are harmful for our environment. (11)

A(iv) An abiotic component of ecosystem is temperature. (11)

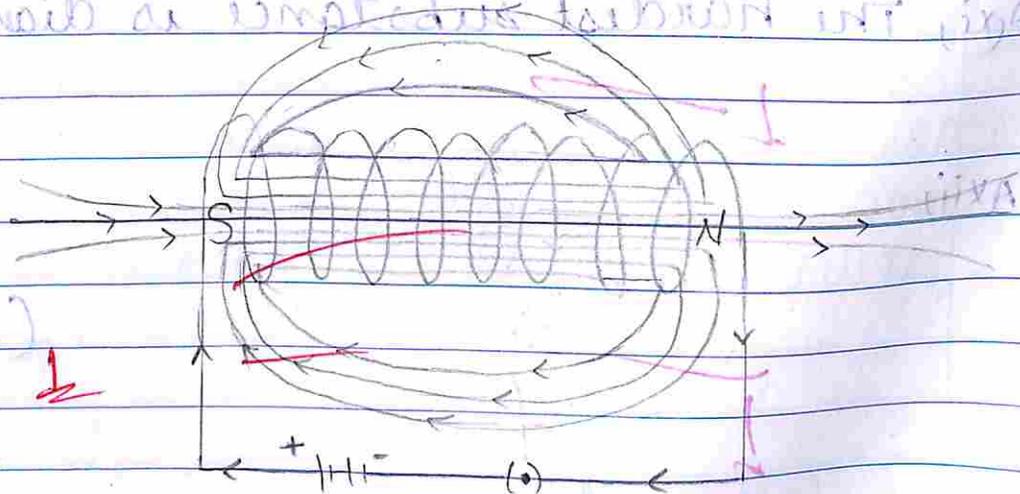
A(v) The value of ~~conductance~~ $P = 2.5 \Omega$ Ans
The value of $G = 16 \times 10^{-4} \Omega m$ Ans

A(vi) Two magnetic field lines do not intersect each other because if we put a compass on the point of intersection then it will the compass



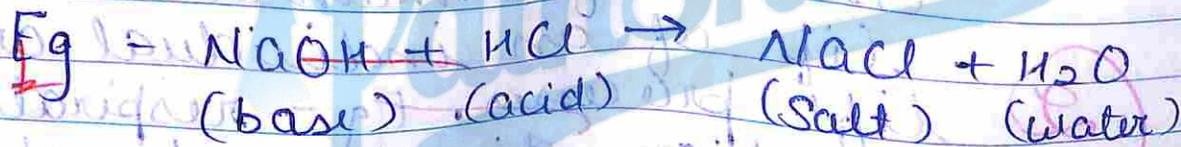
b) will point in two different directions which is not possible.

A(vii)



MAGNETIC FIELD IN A CURRENT CARRYING SOLENOID

A(viii) A reaction in which an acid and a base reacts to form salt and water is called neutralisation reaction.



A(ix) The modern periodic law states that the properties of an element are the periodic function of its atomic number.

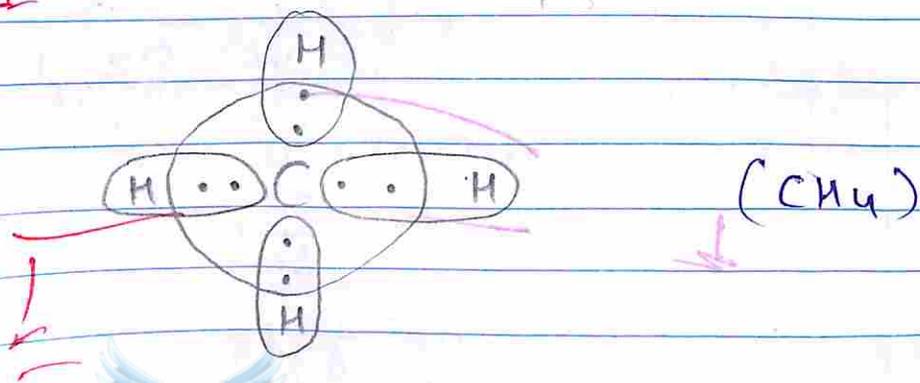
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Axi) In this reaction, CuO is reduced.

Axi) The hardest substance is diamond.

Axii)



This is the electron dot structure of methane.

Section - B

A.4 All the processes which are essential for maintainance of life of an individual are called life processes. Eg - respiration, excretion, digestion, transportation, etc.

A.5 The flower in which either male or female sexual part ~~is~~ present is called unisexual.

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an unisexual flower.
Eg - Papaya and watermelon

A.6



GERMINATION OF POLLEN GRAIN ON STIGMA

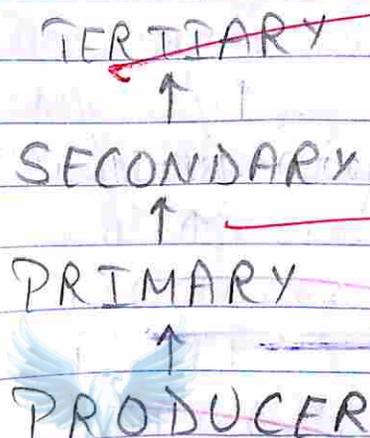
A.7 The autotrophs or producers take sunlight and convert it into chemical energy. This energy is then taken by primary consumer or herbivores. Then secondary consumer or small carnivores take this energy. And finally from secondary consumers this energy is taken by



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Large ~~can~~ carnivores are tertiary consumers. In this way energy flows between the various components of environment. In each step ~~of~~ 10% of energy is lost. J.A



A.8. (i) The ratio of image distance to object distance is called of a lens is called magnification of lens. It is represented by letter 'm'. It also tells us about the nature of image.

$$m = \frac{-v \text{ (image distance)}}{u \text{ (object distance)}}$$

(ii) Convex mirror is used as rear-view mirror in vehicles.



A.9 (i), The electric potential difference in a current carrying conductor can be defined as the work done to move a charge from one point to another of the conductor. The S.I. unit of potential difference is volt. And it is represented by 'V'.

A.10 (ii), Electric heater, electric toaster are the devices which are based on Joule's law of heating.

A.10 (i) Let the current I be flowing through each resistor (R_1, R_2, R_3). The potential difference across the current carrying conductor is V . Let V_1, V_2, V_3 be the potential difference of resistors R_1, R_2, R_3 respectively. Let R be the total resistance.

$$V = V_1 + V_2 + V_3 \quad \text{--- (i)}$$

By Ohm's Law

$$I = \frac{V}{R} \quad V = IR$$

So, $V_1 = IR_1$ --- (ii)

$$V_2 = IR_2 \quad \text{--- (iii)}$$



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$V_3 = IR_3$ (iii), (iv)

By eq. (i), (ii), (iii) and (iv) we get,

$IR = IR_1 + IR_2 + IR_3$

$VR = V(R_1 + R_2 + R_3)$

$R = R_1 + R_2 + R_3$

A.11 The household electric circuits, the electric appliances are connected in parallel combination because -

2

(i) all appliances require different value of current to operate.

(ii) In case one appliance is damaged, then, other appliances will not be affected by this.

A.12 According to this rule, stretch your forefinger, middle finger and thumb of left hand in such a manner that they are mutually perpendicular. If the forefinger points in the direction of magnetic field and the middle finger points in the

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direction of current then the thumb will point in the direction of force acting on the conductor.

A(ii) When the magnetic field is changed then the current flowing in one conductor will induce in another conductor. This is called electromagnetic induction.

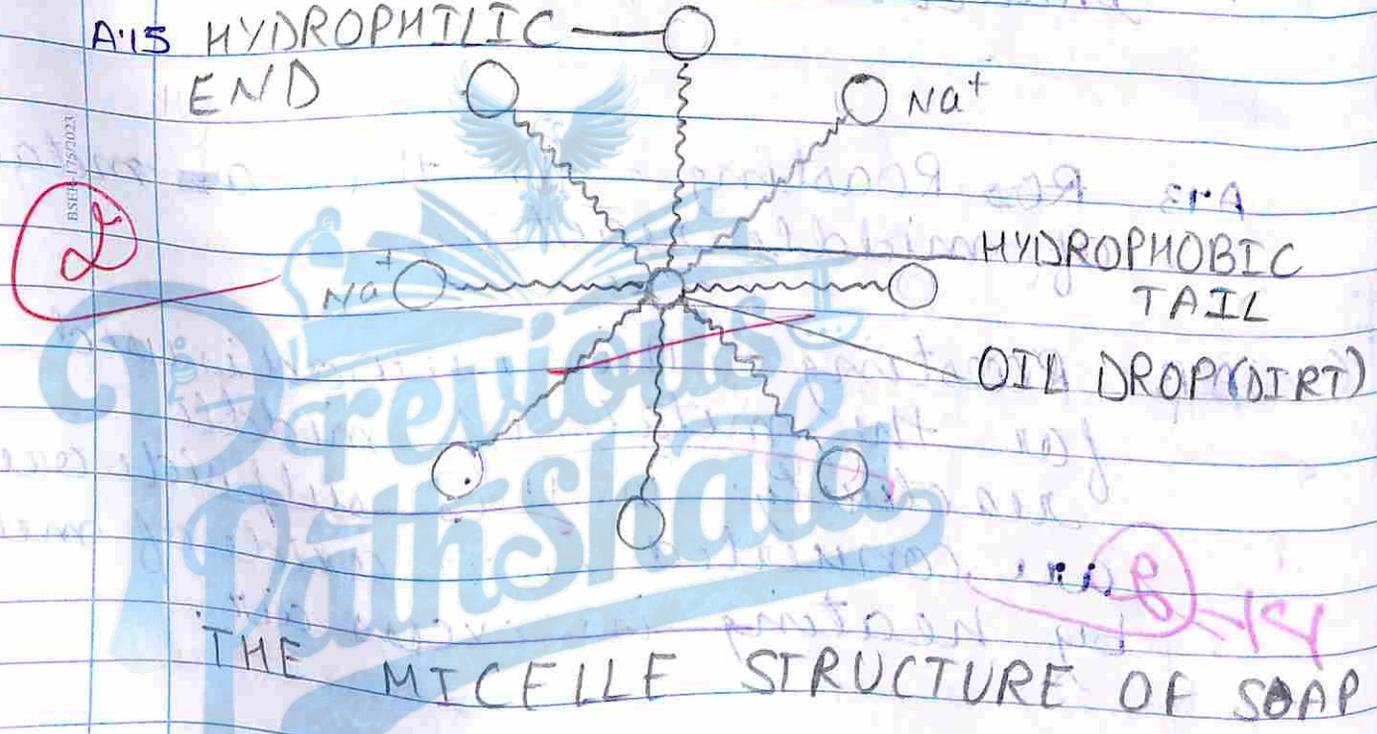
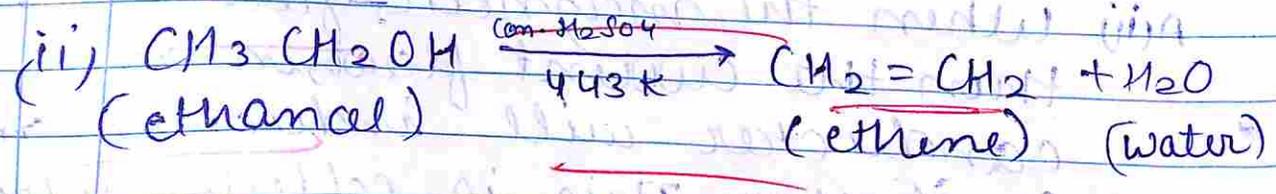
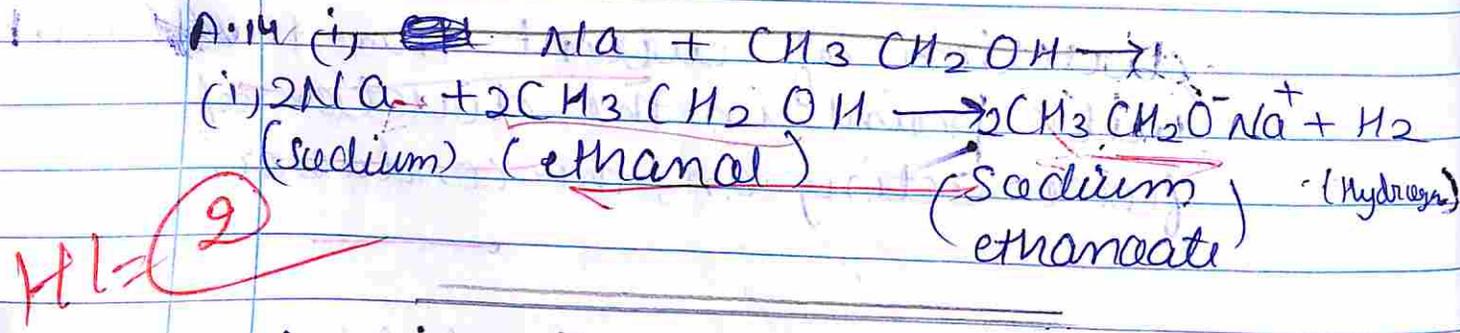
A13 ~~Ro~~ Roasting - In this a metal of middle reactivity

A13 ~~Roasting~~ Roasting - This method is used for the metals of middle reactivity. In this sulphide ores are converted into oxide of metal by heating in excess air.

Calcination - This method is also used for the metals of middle reactivity. In this carbonate ores are converted into oxides of metal by heating in limited air.



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A.16 Newland arranged the known elements according to their physical and chemical properties and in increasing atomic masses. He started from He observed

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that every eighth element has properties similar to that of first. He compared this to the octaves of music. This is known as Newland's law of octaves.

A.16 According to Newland's law of octaves, ~~every~~ elements

9 when arranged in increasing atomic mass, every eighth element has properties similar to that of first.

Section - C

A.17 The hormone secreted by ovaries is oestrogen. Its functions are-

- (i) To regulate menstrual cycle.
- (ii) To develop female sex organ.

A.18 Homologous organs - These organs

12+12 which have similar origin and structure but have different functions. Eg - limbs of lizard and frog.



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~~2~~ Analogous organs - These organs ^{which have} different origin and structure but perform same functions.
Eg - wings of bat and bird.

A:19 (i) Stars & Light

A:19 (i) When the light of stars enter in first layer of atmosphere, it bends as it enters the next layer. And when it enters our eyes this process is continued till the light reaches our eyes. This changes the apparent position of star. ~~And~~ As an

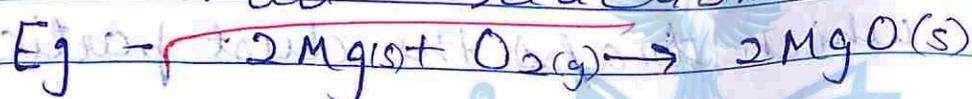
A:19 (i) The different layers of atmosphere are multilayered. But their temperature and density changes continuously. Due to this the apparent position of star also changes continuously. Thus the continuous change in apparent position of stars ~~leads~~ leads to the twinkling of star.

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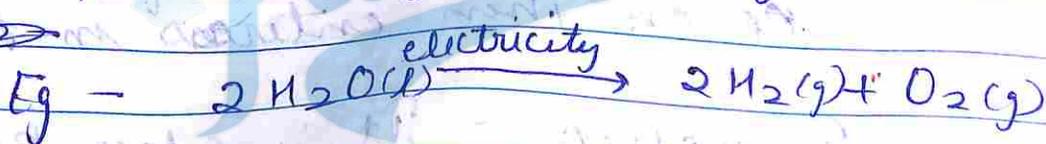
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Aii) The danger signals are red in colour because the red colour is the ~~least scattered~~ by fog or smoke. Hence, it can be easily seen from ~~far away~~ distances.

A:20 (i) Combination reactions - These reactions in which two or more reactants react to form a ~~single~~ single product are called combination reactions.



(ii) Decomposition reaction - These reactions in which a single reactant react to form two or more products. This can be done by heat, electricity and sunlight.



Section - D

A.21 (i) The respiration in which oxygen is inhaled and pyruvate is broken down to give carbon dioxide, water and energy.

1+2+1 = 4

(ii) Nostrils - The oxygen rich air is inhaled through nostrils. It contains hairs which filter the air. It also contains mucus which traps the dust particles.

Pharynx - It is the junction of nasal and buccal cavity. The air from nostrils comes here.

Larynx - It is called voice box. The oxygen rich air from pharynx then enters in larynx.

Trachea - It is also known as wind pipe. Then the air enters the trachea. The trachea is protected by C-shaped cartilage.



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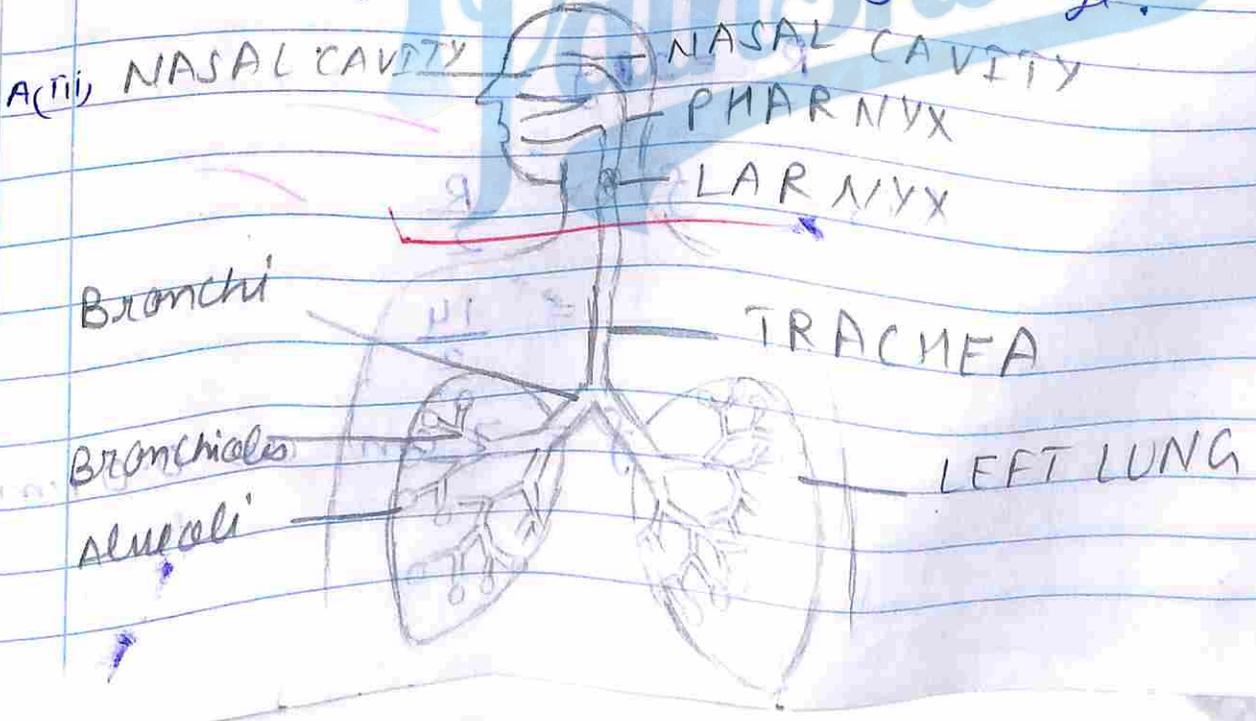
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Bronchi - The trachea is then divided into two main stream called Bronchi. Each bronchi enters ~~to~~ a lung.

Bronchioles - The bronchi ~~to~~ are divided into multiple small branches called bronchioles.

Alveoli - The oxygen rich air from bronchioles enters the alveoli. Alveoli are ~~small sacs~~ which ~~are~~ have balloon like structure. Alveoli are one cell thick. The walls of alveoli are one ~~to~~ cell thick. These walls are highly supplied with blood capillaries. The surface area of Alveoli is very large.



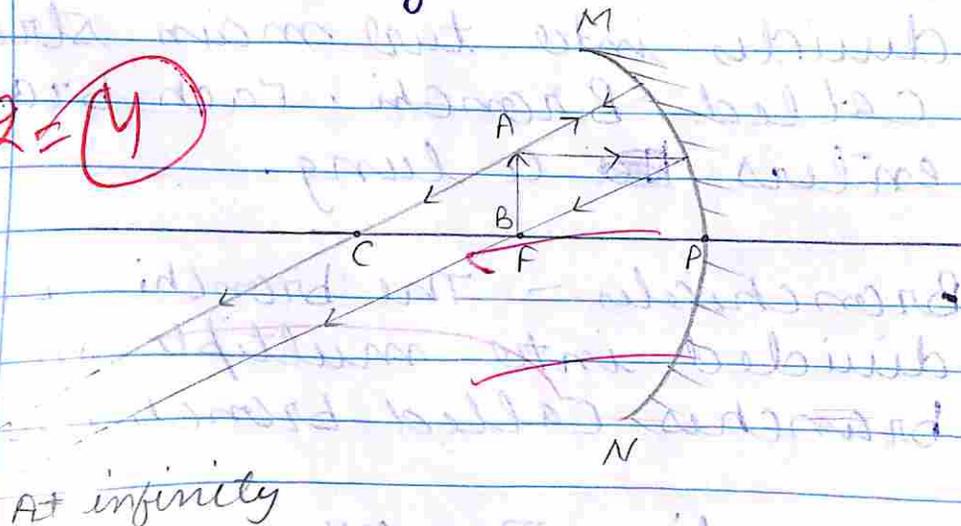


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A.22 (ii) When object is at focus?

$$2 + 2 = 4$$



At infinity

Position of image = At infinity

Size of image = Highly enlarged

Nature of image = Real and inverted

(ii) Radius of curvature = 14 cm,
Focal length of mirror = ?

We know that,

$$R = 2f$$

So, $f = \frac{R}{2}$

$$= \frac{14}{2}$$

$$f = 7 \text{ cm or } 0.07 \text{ m Ans}$$



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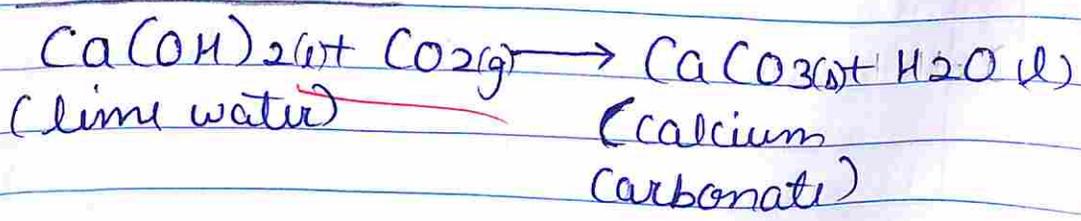
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A:23 (i) Example of a factory indicator is ~~is~~ lime water.

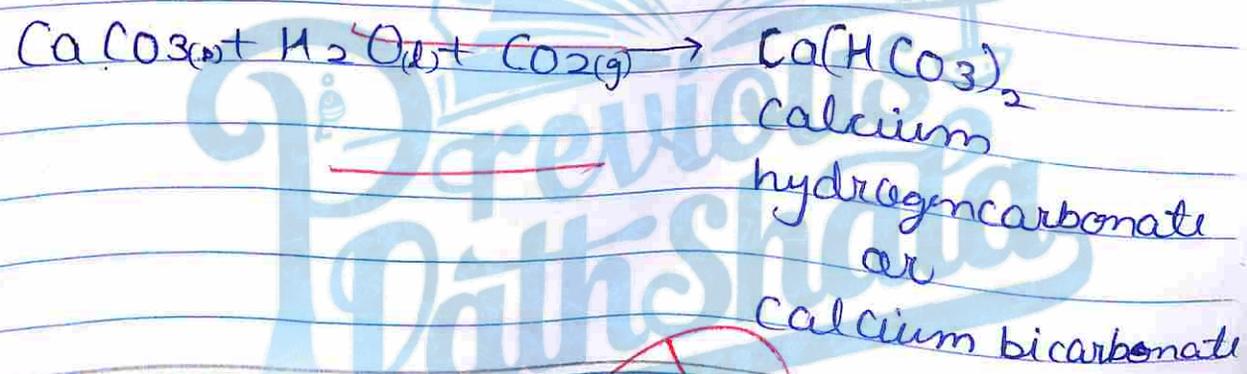
H/H/H
A:ii

Compound [A] = $CaCO_3$
Compound [B] = $Ca(HCO_3)_2$

Equation - 1



Equation - 2



total \Rightarrow 173
guy